

12

**EUROPEAN PATENT APPLICATION**

21 Application number: 85309450.6

51 Int. Cl.<sup>4</sup>: **C 07 F 7/08**  
**C 08 G 77/38**

22 Date of filing: 23.12.85

30 Priority: 22.12.84 JP 271506/84

43 Date of publication of application:  
02.07.86 Bulletin 86/27

64 Designated Contracting States:  
DE FR GB

71 Applicant: Chisso Corporation  
8-32, Nakanoshima 3-chome Kita-ku  
Osaka-shi Osaka-fu(JP)

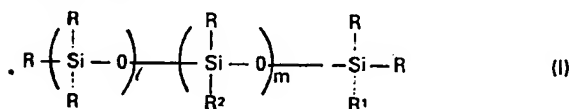
72 Inventor: Nakano, Takaharu  
523, Kubiri 2-chome  
Yokosukashi Kanagawaken(JP)

72 Inventor: Ohtake, Nobumasa  
10-2, Otsutomocho Kanazawaku  
Yokohamashi Kanagawaken(JP)

74 Representative: Lamb, John Baxter et al,  
Marks & Clerk 57/60 Lincoln's Inn Fields  
London WC2A 3LS(GB)

54 Carboxyl group-containing siloxane compounds.

57 Carboxyl group-containing siloxane compounds have the formula:



in which R is a C<sub>1</sub>C<sub>4</sub> alkyl group; R<sup>1</sup> is a group R or a group R<sup>2</sup>; R<sup>2</sup> is a group CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>(OCH<sub>2</sub>CH<sub>2</sub>)<sub>n</sub>COOH (in which n is an integer of 1 or more); l is 0 or an integer of 1 or more; m is 0 or an integer of 1 or more; l + m is an integer of 1 or more; and R<sup>1</sup> is a group R<sup>2</sup> when m is 0.

M&amp;C FOLIO: 799P51207

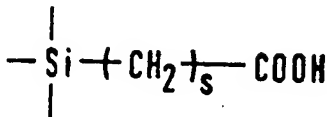
WANGDOC: 02961

CARBOXYL GROUP-CONTAINING SILOXANE COMPOUNDS

This invention relates to carboxyl group-containing siloxane compounds.

In general, carboxyl group-containing siloxane compounds are useful for a number of applications where, for example, organosiloxane compounds soluble in water or alcohols are desired. Thus, for example, carboxyl group-containing siloxane compounds may be used as emulsifying agents for forming aqueous emulsions of organosiloxane polymers or may be used in alcohol-based cosmetics. Further, carboxy group-containing siloxane compounds may be expected to have strong adhesion onto inorganic materials and to modify the surface of the materials so as to impart to the surface functions such as water repellency, stain resistance, non-adhesive properties, heat resistance, abrasion resistance, etc. For example, as disclosed in Japanese laid-open patent applications Nos. Sho 53-10882/1978 or Sho 57-10145/1982, such siloxane compounds have been used as an ink-repelling material for litho printing.

In a conventional carboxyl group-containing siloxane compound the carboxyl group is linked to the compound by a group of the formula:

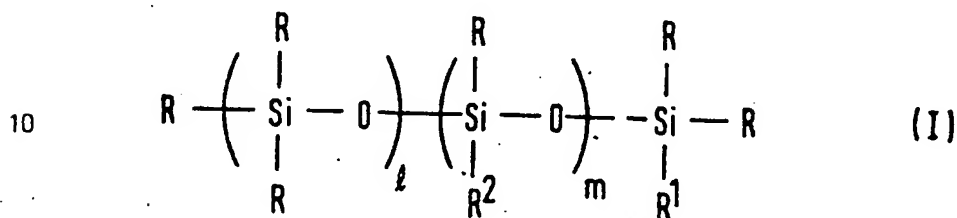


5 wherein g is 2, 3 or 4, as described, for example, in Japanese patent publications Nos. Sho 40-20279/1965, Sho 41-236/1966, Sho 42-6519/1967 and Sho 49-4840/1974. We have found that such conventional carboxyl group-containing siloxane compounds are stable to heat when  
10 they have a number average molecular weight (Mn) greater than 1,000, but are decomposed by heat when Mn is 1,000 or less.

We have further found that when a carboxyl group is bonded to the compound via a difunctional group  
15 containing a polyoxyethylene chain, the resultant compounds are stable to heat even if they are of low molecular weight.

The object of the present invention is, thus, to provide a carboxyl group-containing siloxane compound, whether of low or high molecular weight, having a superior heat stability and which may be used as an emulsifying agent, a surface modifier for inorganic materials, etc.

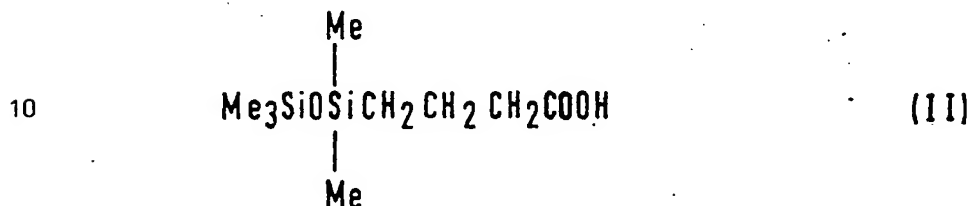
According to the invention there is provided a carboxyl group-containing siloxane compound of the formula



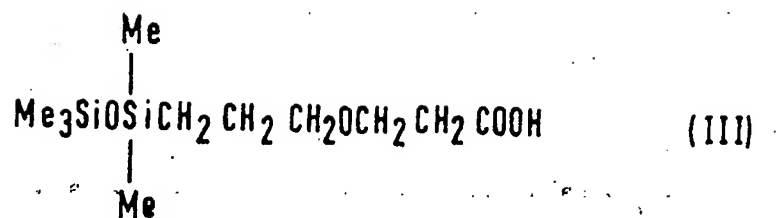
in which R is a C<sub>1</sub>-C<sub>4</sub> alkyl group; R<sup>1</sup> is a group R or a group R<sup>2</sup>; R<sup>2</sup> is a group -CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>- (OCH<sub>2</sub>CH<sub>2</sub>)<sub>n</sub> COOH (in which n is an integer of 1 or more); l is 0 or an integer of 1 or more; m is 0 or an integer of 1 or more; l+m is an integer of 1 or more; and R<sup>1</sup> is a group R<sup>2</sup> when m is zero.

In the above general formula,  $\underline{n}$  has no particular upper limit and may be e.g. several thousands or several tens of thousands. The  $\underline{m}$  and  $\underline{l}$  also each have no particular upper limit and may be e.g. several millions or several tens of millions.

The difference in behaviour between the compounds of the inventions and conventional carboxyl group-containing siloxane compounds is illustrated by the fact that a compound of the formula:

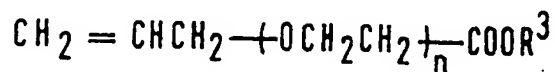


(in which Me is a methyl group) undergoes a ring closure reaction at about 150°C to give a product of unknown structure, whereas a compound of the formula



(which is in principle the compound of formula (II) containing an added ethyleneoxy group) is sufficiently stable to permit of distillation at 150°C/1 mm Hg.

5 The carboxyl group-containing compounds of the invention may be prepared by subjecting an ester of the formula:

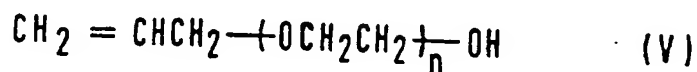


(in which R<sup>3</sup> is, for example, a methyl, ethyl or  
 10 trimethylsilyl group) and an Si-H-containing siloxane compound to hydrosilylation and subsequently hydrolysing the resulting ester.

The esters of formula (IV) compound in which  $n$  is 1 and  $R^3$  is methyl or ethyl can be easily obtained by subjecting allyl alcohol to an addition reaction with acrylonitrile in the presence of a basic catalyst [see 5 Ind. Eng. Chem., 44, 2867 (1952)], followed by alcoholysis of the nitrile group of the resulting product. When methanol is used as a solvent in the alcoholysis, the methyl ester can be obtained, while when ethanol is used, the ethyl ester can be obtained 10 [see Org. Synth., 1, 270 (1941)]. Further, when the esters obtained by the above reactions are further hydrolyzed in the presence of a basic catalyst to obtain the corresponding carboxylic acid, followed by reaction this acid with hexamethyldisilazane, it is possible to 15 obtain the trimethylsilyl ester [see J. Org. Chem., 40, 1610 (1975)].

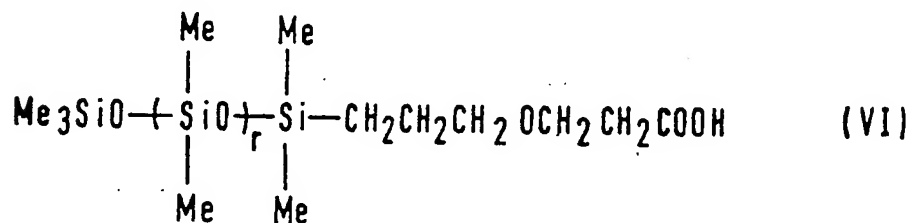
Esters of formula (IV) in which  $n$  is 2 or more can be prepared in a similar manner using in place of the allyl alcohol, a compound of the formula:

20



which may be prepared by the addition reaction of the necessary number of mols of ethylene oxide with allyl alcohol.

A specific class of carboxyl group-containing siloxane compounds of the invention are those of the formula:



wherein  $r$  represents 0 or an integer of 1 or more.

The compound (VI) may be readily obtained from (a),  
 10 when  $r$  is 0, a readily commercially available siloxane compound containing a hydrogen atom at one end or (b)  
 when  $r$  is 1 or more, siloxane compounds containing a hydrogen atom at one end obtained by reacting lithium trimethylsiloxanolate with hexamethylcyclotrisiloxane to  
 15 give a siloxane compound containing H atom at its one end [see Polym. Preprints, 10 (2), 1361 (1969)], and then subjecting these to addition reaction with an ester of formula (IV) in which  $R^3$  is a trimethylsilyl group in the presence of a catalyst for addition reaction.

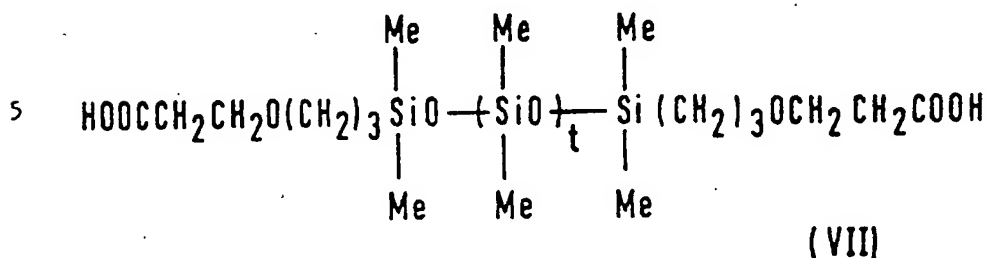


preferably under a nitrogen gas atmosphere, followed by  
subjecting the resulting product to detrimethyl-  
silylation with an alcohol.

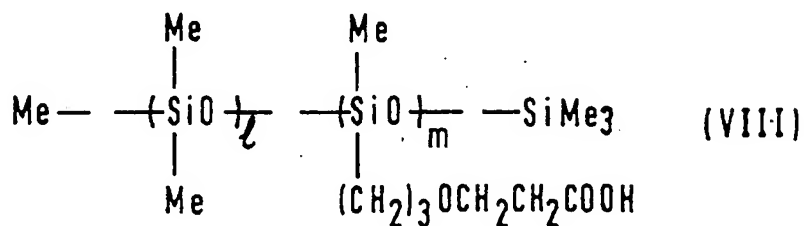
As to the siloxane compound containing an H atom at  
5 one end, it is possible to optionally prepare a siloxane  
compound having a controlled molecular weight and  
molecular weight distribution, as far as its average  
molecular weight is about 10,000 or lower. In the  
reaction of the siloxane compound having an H atom at  
10 one end which the trimethylsilyl ester, the ester is  
suitably used in at least equimolecular quantity to the  
siloxane compound, preferably in an amount of 1.2 moles  
per mole of siloxane compound. The reaction temperature  
is suitably from 40 to 200°C, preferably from 80 to  
15 130°C. Suitable catalysts for the addition reaction are  
complex compounds of metals of group 8 of the Periodic  
Table may be exemplified, e.g. platinum compounds,  
rhodium compounds or palladium compounds such as known  
alcohol compounds, aldehyde compounds or the like of  
20 chloroplatinic acid, complexes of chloroplatinic acid  
with various olefins, etc.

The alcohol used for detrimethylsilylation is  
preferably methanol or ethanol.

The following carboxyl group-containing siloxane compounds (VII) and (VIII) can be easily prepared by similarly reacting the corresponding SiH-containing compound with the trimethylsilyl ester.



wherein  $t$  is 0 or an integer of 1 or more.



(In case of this example  $m$  is an integer of 1 or more.)

The compounds of the invention are useful for various applications, for example as emulsifying agents for forming aqueous emulsions of organosiloxane polymers, as ingredients for alcohol-based cosmetics, or  
5 as a surface modifier for imparting to the surface of inorganic materials, functions such as water repellency, stain resistance, non-adhesive properties, heat resistance, abrasion resistance, etc.

In order that the invention may be well understood  
10 the following examples are given by way of illustration only.

Reference example

- (1) Allyl alcohol (150 g, 2.58 mols) and an ion exchange resin (IRA-400, trade name of a strongly basic anion exchange resin made by Rohm & Haas Company, U.S.A.) (25 g) were fed into a flask in  $N_2$  current and the temperature was kept at  $45^\circ C$ , followed by dropwise adding acrylonitrile (125 g, 2.35 mols) over about 1 to 2 hours, thereafter agitating the mixture at  $45^\circ C$  for about 8 to 9 hours, filtering off the resin and carrying out vacuum distillation to obtain cyanoethyl allyl ether (196 g,  $93 \sim 96^\circ C/20$  mmHg). Yield: 75%.
- (2) Ethanol (350 ml), water (34.4 ml) and conc. sulfuric acid (200 ml) were fed into a flask, followed by dropwise adding cyanoethyl allyl ether (222 g, 2 mols) at room temperature over 30 minutes, thereafter agitating the mixture at a reaction temperature of  $100 \sim 110^\circ C$  for about 7 hours, pouring the reaction fluid into water, extracting it with isopropyl ether, washing the extract solution with 5%  $NaHCO_3$  aqueous solution till the extract solution became neutral, drying over  $MgSO_4$ , and subjecting the extract solution to vacuum distillation to obtain 2-allyloxypropionic acid ethyl ester (147.2 g,  $94^\circ C/18$  mmHg). Yield: 46%.
- (3) 2-Allyloxypropionic acid ethyl ester (147.2 g, 0.93 mol), water (300 ml) and NaOH (44.7 g) were fed

into a flask, followed by agitating the mixture for about 5 to 6 hours while the reaction temperature was kept at about 60°C, thereafter dropwise adding conc. hydrochloric acid (90 ml) under ice cooling, extracting  
5 the resulting deposited oily substance with isopropyl ether, drying over  $\text{MgSO}_4$  and carrying out vacuum distillation to obtain 2-allyloxypropionic acid (95.7 g, 108°C/5 mmHg). Yield: 79%.

The thus obtained 2-allyloxypropionic acid (170.2 g, 1.31 mol) was fed into a flask, followed by dropwise adding  
10 hexamethyldisiloxane (128.8 g, 0.8 mol) in  $\text{N}_2$  current at room temperature over one hour, thereafter raising the reaction temperature to 80°C, then agitating the mixture for about 3 hours, and carrying out vacuum  
15 distillation to obtain 2-allyloxypropionic acid trimethylsilyl ester (3c) (231.4 g). Yield: 87.5%.

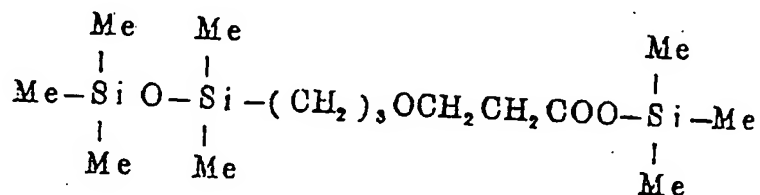
#### Example 1

2-Allyloxypropionic acid trimethylsilyl ester (65.5 g, 0.32 mol) obtained in the above Reference example, and  
20 a solution of chloroplatinic acid in isopropanol (0.042 mol; chloroplatinic acid 1 g/20 ml) were fed into a flask in  $\text{N}_2$  current, followed by raising the temperature to 100°C, thereafter dropwise adding pentamethyldisiloxane (40 g, 0.27 mol) with stirring over 30 minutes, then further  
25 carrying out reaction at 100°C for 2 hours, and subjecting the resulting reaction mixture solution to vacuum

distillation to obtain a colorless, transparent liquid having a b.p. of 120°C/1 mmHg (77.9 g, yield 82.4%).

This product was confirmed to be a silicon compound having the following structural formula, from the

5 following analytical results:



H-NMR(C<sub>2</sub>H<sub>4</sub>):

δ 0.06 (Si-CH<sub>3</sub>, s, 15H)

0.3 (-CO<sub>2</sub>SiMe<sub>3</sub>, s, 9H)

0.56 (-CH<sub>2</sub>-Si, m, 2H)

1.56 (-CH<sub>2</sub>-, m, 2H)

2.5 (-CH<sub>2</sub>-, t, 2H, J = 6 Hz)

3.3 (-CH<sub>2</sub>-, t, 2H, J = 6 Hz)

3.6 (-CH<sub>2</sub>-, t, 2H, J = 6 Hz)

IR(KBr):

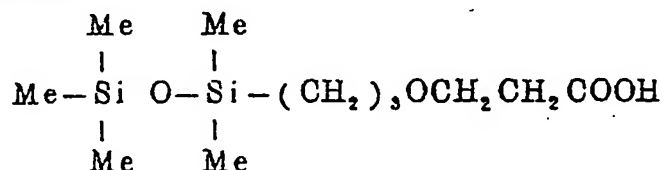
ν<sub>max</sub> 2960 cm<sup>-1</sup> (C-H)

1740 cm<sup>-1</sup> (C=O)

1120~1050 cm<sup>-1</sup> (Si-O)

MSm/e: 350 (M<sup>+</sup>)

Next, the thus obtained compound (77.9 g, 0.23 mol) and methanol (100 ml) were fed into a flask, followed by agitating the mixture at room temperature for about 2 ~ 3 hours and subjecting the resulting reaction fluid to vacuum distillation to obtain a colorless, transparent liquid having a b.p. of 150°C/1 mmHg (61 g, yield: quantitative). This product was confirmed to be an oxyethylenecarboxylic acid-modified compound having the following structural formula, from the following analytical results.



$^1\text{H-NMR}(\text{C}_2\text{Cl}_4)$ :

$\delta$  0.06 (Si-CH<sub>3</sub>, s, 15H)  
 0.56 (-CH<sub>2</sub>-Si, m, 2H)  
 1.6 (-CH<sub>2</sub>-, m, 2H)  
 2.5 (-CH<sub>2</sub>-, t, 2H, J = 6 Hz)  
 3.3 (-CH<sub>2</sub>-, t, 2H, J = 6 Hz)  
 3.6 (-CH<sub>2</sub>-, t, 2H, J = 6 Hz)  
 11.6 (-CO<sub>2</sub>H, s, 1H)

IR(KBr):  $\nu_{\text{max}}$  3050 (CO<sub>2</sub>H)

2960  $\text{cm}^{-1}$  (C-H)

1740  $\text{cm}^{-1}$  (C=O)

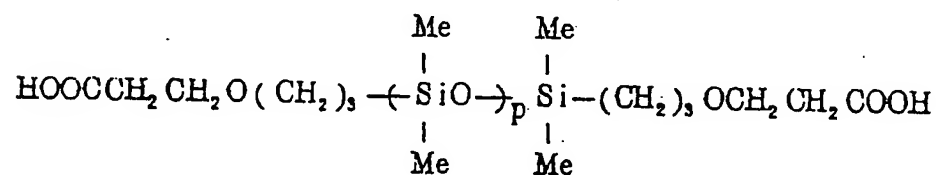
1120~1050  $\text{cm}^{-1}$  (Si-O)

MSm/e: 278 (M<sup>+</sup>)

This product was stable without causing any ring closure reaction in the vicinity of 150°C.

Example 2

2-Allyloxypropionic acid trimethyl ester (40.2 g,  
 5 0.2 mol) was reacted with a siloxane polymer containing  
 H atom at both the ends thereof ( $\bar{M}_n$  1,100, H equivalent  
 547) in the same manner as in Example 1, followed by  
 distilling off unreacted raw materials and a low boiling  
 substance from the reaction mixture fluid under a reduced  
 10 pressure (150°C/1 mmHg) for 2 hours, and washing the  
 residual fluid with water till the residual fluid  
 became neutral to obtain a colorless, transparent liquid  
 (111.6 g). This product was confirmed to be an oxyethylene-  
 carboxylic acid-modified silicone having the following  
 15 structural formula, from the following analytical results:



IR(KBr) :  $\nu_{\text{max}}$  3100 (CO<sub>2</sub>H)

3000~2950 (C-H)

1740 (C=O)

1125~1060 (Si-O)



Carboxylic acid equivalent: 672 (theoretical  
value 677)

$\bar{M}_n$ : 1344 (calculated from carboxylic acid equivalent)

Side chain signals were confirmed according to H-NMR:

$\delta$  0.56 ( $-\text{CH}_2-\text{Si}$ , m, 2H)

1.6 ( $-\text{CH}_2-$ , m, 2H)

2.5 ( $-\text{CH}_2-$ , t, 2H,  $J = 6 \text{ Hz}$ )

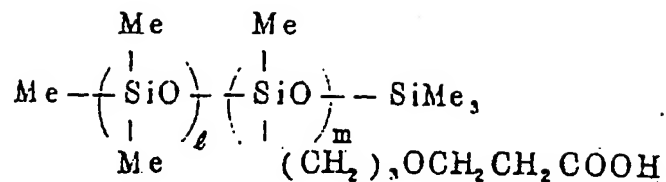
3.3 ( $-\text{CH}_2-$ , t, 2H,  $J = 6 \text{ Hz}$ )

3.3 ( $-\text{CH}_2-$ , t, 2H,  $J = 6 \text{ Hz}$ )

10 Example 3

Reaction was carried out in the same manner as in Example 2 except that the above siloxane polymer containing H atom at both the ends thereof was replaced by a siloxane polymer containing pendant type H atom

15 (41.9 g,  $\bar{M}_n$  5300, H equivalent 252) to obtain a colorless, transparent liquid (61.6 g). This product was confirmed to be an oxyethylenecarboxylic acid-modified silicone having the following structural formula, from the following analytical results:



IR (KBr):  $\nu_{\max}$  3110 (CO<sub>2</sub>H)

3000~2950 (C-H)

1740 (C=O)

1125~1050 (Si-O)

Carboxylic acid equivalent: 390 (theoretical  
value 382)

$\bar{M}_n$ :  $\approx$  7800 (calculated from GPC)

$\bar{P}$ :  $\approx$  55,  $\bar{m}$ :  $\approx$  20 (calculated from GPC, H equivalent and infrared absorption spectra)

Side chain signals were confirmed according to H-NMR:

$\delta$  0.56 (-CH<sub>2</sub>-Si, m, 2H)

1.6 (-CH<sub>2</sub>-, m, 2H)

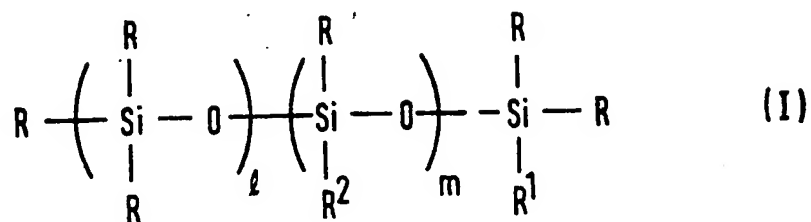
2.5 (-CH<sub>2</sub>-, t, 2H, J = 6 Hz)

3.3 (-CH<sub>2</sub>-, t, 2H, J = 6 Hz)

3.3 (-CH<sub>2</sub>-, t, 2H, J = 6 Hz)

Claims

1. A carboxyl group-containing siloxane compound of the formula



in which R is a C<sub>1</sub>-C<sub>4</sub> alkyl group; R<sup>1</sup> is a group R or group R<sup>2</sup>; R<sup>2</sup> is a group -CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub> (OCH<sub>2</sub>CH<sub>2</sub>)<sub>n</sub> COOH (in which n is 0 or an integer of 1 or more); l is 0 or an integer of 1 or more; m is 0 or an integer of 1 or more; l+m is an integer of 1 or more; and R<sup>1</sup> is a group R<sup>2</sup> when m is 0.

2. A compound according to claim 1 characterised in that m is ten million or less.

3. A compound according to claim 2 characterised in that m is one million or less.

4. A compound according to any one of the preceding claims characterised in that l is ten million or less.

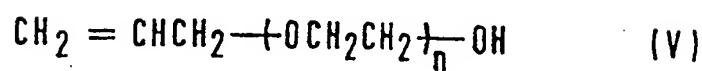
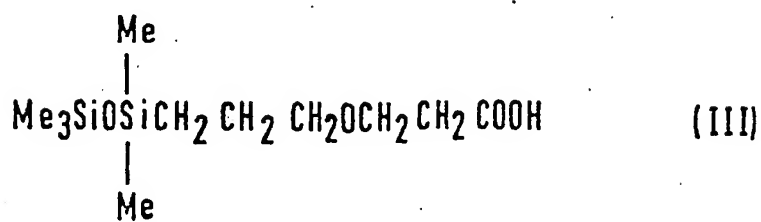
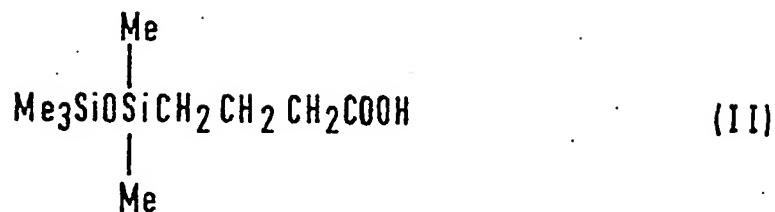
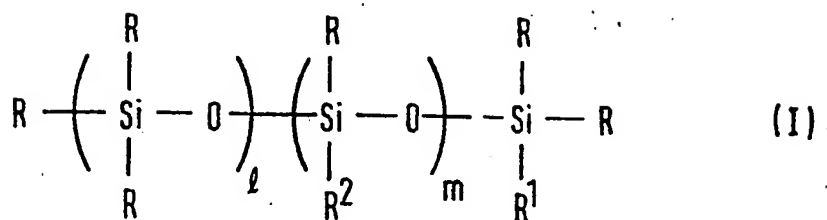
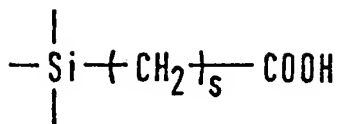
5. A compound according to claim 4 characterised in that l is one million or less.

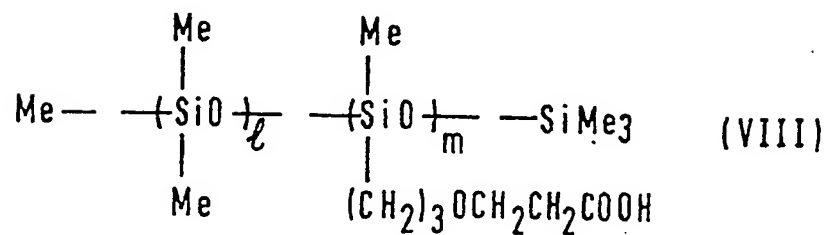
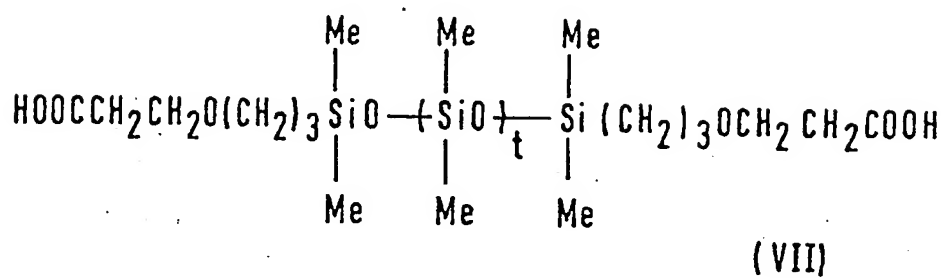
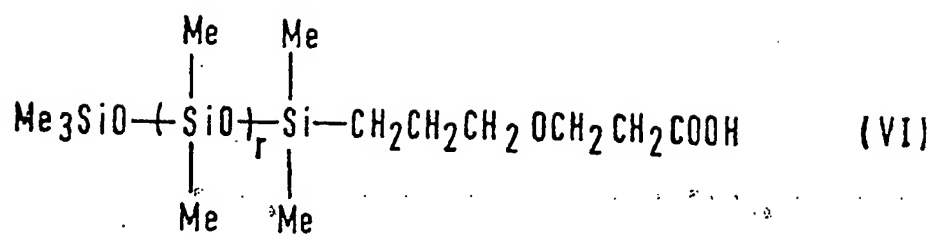
6. A compound according to any one of the preceding claims characterised in that n is 10,000 or less.

7. A compound according to claim 6 characterised in that n is 1,000 or less.

8. A compound according to claim 1 characterised in that it has a number average molecular weight of 7,8000 or less.

9. A compounds according to claim 8 characterised in that it has a number average molecular weight of 1,000 or less.





⑫

**EUROPEAN PATENT APPLICATION**

⑰ Application number: 85309450.6

⑤① Int. Cl.<sup>4</sup>: **C 07 F 7/08, C 08 G 77/38**

⑳ Date of filing: 23.12.85

③① Priority: 22.12.84 JP 271506/84

⑦① Applicant: **Chisso Corporation, 6-32, Nakanoshima 3-chome Kita-ku, Osaka-shi Osaka-fu (JP)**

④③ Date of publication of application: 02.07.86  
Bulletin 86/27

⑦② Inventor: **Nakano, Takaharu, 523, Kubiri 2-chome, Yokosukashi Kanagawaken (JP)**  
Inventor: **Ohtake, Nobumasa, 10-2, Otsutomocho Kanazawaku, Yokohamashi Kanagawaken (JP)**

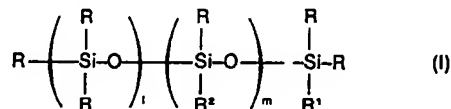
⑧④ Designated Contracting States: **DE FR GB**

⑧⑧ Date of deferred publication of search report: 11.11.87 Bulletin 87/46

⑦④ Representative: **Lamb, John Baxter et al, MARKS & CLERK 57/60 Lincoln's Inn Fields, London WC2A 3LS (GB)**

⑤④ **Carboxyl group-containing siloxane compounds.**

⑤⑦ Carboxyl group-containing siloxane compounds have the formula:



In which R is a C<sub>1</sub>C<sub>4</sub> alkyl group; R<sup>1</sup> is a group R or a group R<sup>2</sup>; R<sup>2</sup> is a group CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub> (OCH<sub>2</sub>CH<sub>2</sub>)<sub>n</sub> COOH (in which n is an integer of 1 or more); l is 0 or an integer of 1 or more; m is 0 or an integer of 1 or more; l + m is an integer of 1 or more; and R<sup>1</sup> is a group R<sup>2</sup> when m is 0.

**EP 0 186 507 A3**



European Patent  
Office

# EUROPEAN SEARCH REPORT

0186507

Application number

EP 85 30 9450

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.4)
X	CHEMICAL ABSTRACTS, vol. 54, no. 8, 25 April 1960, column 7534a-c, Columbus, Ohio, US; A.D. PETROV et al.: "Synthesis of alpha,omega-bis(cyanoalkyl)tetraalkylsiloxanes", & Doklady Akad. Nauk S.S.S.R. 128, 1204-7(1959) * Whole document *	1-9	C 07 F 7/08 C 08 G 77/38
X	DE-A-1 795 082 (I.C.I.) * Pages 1,2; examples *	1-9	
A	US-A-4 139 546 (BERGER) * Claims *	1-9	
D, A	CHEMICAL ABSTRACTS, vol. 97, no. 6, 9 August 1982, page 537, abstract no. 47169e, Columbus, Ohio, US; & JP-A-82 10 145 (TOYOBO CO., LTD) 19-01-1982 * Whole document *	1-9	TECHNICAL FIELDS SEARCHED (Int. Cl.4) C 07 F 7/00 C 08 G 77/00
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 20-08-1987	Examiner PAUWELS G.R.A.
<b>CATEGORY OF CITED DOCUMENTS</b> X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document			